

Chemguide – questions

GROUP 4: INSOLUBLE LEAD COMPOUNDS

1. In each case, describe what you would see if you added the named substance to lead(II) nitrate solution, and write the ionic equation(s) for the reaction(s) involved.
 - a) dilute hydrochloric acid;
 - b) dilute sulphuric acid;
 - c) sodium iodide solution;
 - d) sodium hydroxide solution.